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Toxicity of Benzyl Benzoate from *Kaempferia rotunda* L. Rhizome

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Abstract. Benzyl benzoate is the major component of the essential oil of *K. rotunda* L. rhizome. They are potential to be developed as the medicinal compound. However, the toxicity study of benzyl benzoate as the bioactive compound was still limited. Therefore, the toxicity of benzyl benzoate was investigated. The isolation steps include the extraction of *K. rotunda* L. rhizome using acetone by maceration, then acetone extract was partitioned with *n*-hexane and chloroform respectively. The benzyl benzoate from *n*-hexane fraction was isolated using vacuum liquid chromatography and radial chromatography. The molecular structure of benzyl benzoate was determined based on NMR (1D and 2D) spectroscopic data. The toxicity assay of acetone extract and isolated compounds carried out using the brine shrimp lethality test (BSLT). BSLT results were presented through the lethal concentration 50 (LC₅₀). The toxicity evaluation confirms that acetone extract, *n*-hexane fraction of *K. rotunda* L rhizome and the benzyl benzoate have biological activity with LC₅₀ 35.86, 49.80, and 173.49 µg/mL respectively.

INTRODUCTION

Kaempferia rotunda (Zingiberaceae) is a medicinal plant in Indonesia. It was known locally name as “*kunci pepet*” or “*kunir putih*”. The rhizome of *K. rotunda* was used for traditional medicine such as treating stomach pain, fever, accelerate wound healing, carminative and inflammation due to bruises or sprains [1]. The extracts, essential oils and isolated compounds from *K. rotunda* L rhizome exhibited the essential biological activities. The extract of *K. rotunda* rhizome showed antioxidant activity [2, 3], insecticides [4], anti-inflammatory [5], anthelmintic [6] and antimicrobial activities [7,8]. Some compounds of the *K. rotunda* rhizome also some biological activities. A 2-hydroxy-4,4',6-trimethoxy calkon indicated antioxidant activity with IC₅₀ value of 142 µg/mL [2]. Pinostrobin and 5,7-dihydro flavanone were showed anticancer activity against breast cancer cell T47D with IC₅₀ of 59.8 µg/mL and 122.71 µg/mL respectively [9]. Meanwhile, benzyl benzoate revealed insecticidal activity with an LC₅₀ of 5.6 µg/mL on *Spodoptera littoralis* [4].

The essential oil has an important role in the biological activity of the *K. rotunda*, because of it as a significant component. The essential oil of *K. rotunda* rhizome was contained about 75 compounds with two main compounds namely benzyl benzoate (69.7%) and *n*-pentadecane (22.9%) [10]. In different locations, It was also mentioned that

of 20 compounds in the volatile oil of *K. rotunda* was contain benzyl benzoate 30.61% and cyclopropazulen 28.85% [11]. Furthermore, it was reported that essential oils in the *n*-hexane extract of *K. rotunda* rhizome could inhibit the growth of some bacteria [8]. The other plant study reported that *Cinnamomum aureofulvum* essential oil that contained 43.4% benzyl benzoate showed antibacterial activity against *Staphylococcus aureus*, *S. epidermis*, *Pseudomonas aeruginosa* and *P. cepacia* with minimum inhibitory of 1.87 µg/µL [12]. Then *Salvia urmiensis* essential oil contained 60.3% benzyl benzoate showed high activity against *S. epidermis* and *S. cerevisiae* with minimum inhibitory of 9.3 µg/mL [13]. It showed that benzyl benzoate has potential as a bioactive agent because it is the main component of *K. rotunda* essential oil. In this article, we wish to report the isolation of benzyl benzoate as well as toxicity properties.

Nowadays brine shrimp (*Artemia salina*, fairy shrimp or sea monkeys) lethality assay is commonly used to check the cytotoxic effect of bioactive compounds. The brine shrimp lethality test (BSLT) has been developed for toxicity assay of various concentrations of pure compounds and crude plant extracts. Several advantages of this method are rapidness, simplicity and low requirements [14]. This method has been successfully employed as a bioassay guide for cytotoxic activity and antitumor agents [15].

EXPERIMENTAL

Materials and Instruments

The rhizomes of *K. rotunda* L were collected from Yogyakarta, Indonesia. All chemicals used for extractions and chromatography were of technical grade and analytical grade from Merck.

Structure elucidation of the isolated compound was determined based on ¹H- and ¹³C-NMR (nuclear magnetic resonance) spectra were performed on the Agilent DD2 system operating at 500 (¹H) MHz and 125 (¹³C) MHz and GCMS (Gas Chromatography-Mass Spectrometer): GC17A MSQP 5000 Shimazu.

Isolation of Benzyl Benzoate

Powder of *K. rotunda* L rhizomes (500 g) was extracted with acetone (5 L, 2 times) for 2 days at room temperature, then filtrated and evaporated to give acetone extract. The acetone extracts were partitioned with *n*-hexane and methanol, respectively. *n*-Hexane soluble fractions (15 g) was fractionated by using a silica gel column (vacuum liquid chromatography) and eluted step-wise with 150 mL *n*-hexane, 150 mL *n*-hexane: chloroform (8:2, 7:3, 5.5, 3:7, 2:8), 150 mL chloroform, and 150 mL methanol respectively. All fractions were concentrated on a rotary evaporator, then loaded on the TLC plate with eluent *n*-hexane: chloroform (1:1). The fractions having similar R_f values were pooled sub-fractions, afforded fractions A: 5.9 g, B: 0.9 g, C: 0.5 g, D: 0.8 g, and E: 1.1 g. Fractions A was purified by radial chromatography and eluted with *n*-hexane: chloroform (19:1) yield benzyl benzoate (230 mg). The molecule structures of benzyl benzoate were identified by NMR (1D and 2D) spectrometer.

Toxicity Assays

The toxicity assay was carried out using BSLT method [14] on acetone extract, *n*-hexane soluble fraction and isolated compound (benzyl benzoate). These method was begin with hatching the brine shrimp (*A. salina*) eggs for 48 hours incubations in artificial sea water that prepared by diluting 38 g of sea salt in 1.0 L of distilled-water in a glass chamber. During incubation, the hatching chamber was given constant light source and aerator for oxygen supply. After 24 hours the larvae were fed with added yeast solution 0.06% was into the hatching chamber. The nauplii that 48 hours old were used for toxicity assays.

The test solution was prepared by dissolving 10 mg of sample (acetone extract, *n*-hexane soluble fraction and benzyl benzoate) in 1 mL of 10% v/v dimethylsulfoxide (10000 µg/mL). The stock solution was diluted, so that the sample concentrations of 1000 µg/mL, 500 µg/mL, 250 µg/mL, 100 µg/mL and 10 µg/mL. Six test tubes were labeled as 1, 2, 3, 4, 5 and 6. Then 1 mL of samples solution was taken into the test tubes 1-5 that containing 10 nauplii and 1 mL of seawater. Whereas into the test tube 6 the sample solution was replaced with control (dimethylsulfoxide 10% v/v). After 24 hours of incubation, the number of live nauplii in each concentration and control was counted, then the percentage of dead nauplii (%death) was determined. The experiments were performed in triplicate.

$$\% \text{ Death} = \frac{\text{Number of death nauplii}}{\text{number of death nauplii} + \text{number of live nauplii}} \times 100 \quad (1)$$

The median lethal concentration (LC₅₀) of the test samples is obtained by a plot of percentage of the dead shrimps against the logarithm of the sample concentration. LC₅₀ values are estimated using a probit regression analysis, were analyzed with the SPSS 16.0 program for probit analysis to determine LC₅₀ values and 95% confidence intervals.

RESULTS AND DISCUSSION

Identification of Benzyl Benzoate from *K. rotunda* Rhizome

The identification of isolated compounds by GCMS and NMR spectrometers (1D and 1D) proves that it is benzyl benzoate (Fig. 1).

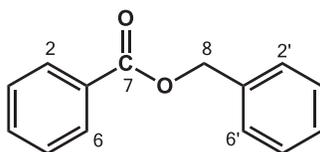


FIGURE 1. Benzyl benzoate

Based on the GCMS analysis, the GC chromatogram shows one main peak with a retention time of 28.146 minutes and an area of 100%. These show that the isolated compound is quite pure. The MS spectra show that the molecular ion peak at m/z 212 [M⁺] is attributed to the molecular mass of benzyl benzoate (C₁₄H₁₂O₂). The GC-MS spectra of the isolated compound are presented in Fig. 2 and Fig. 3.

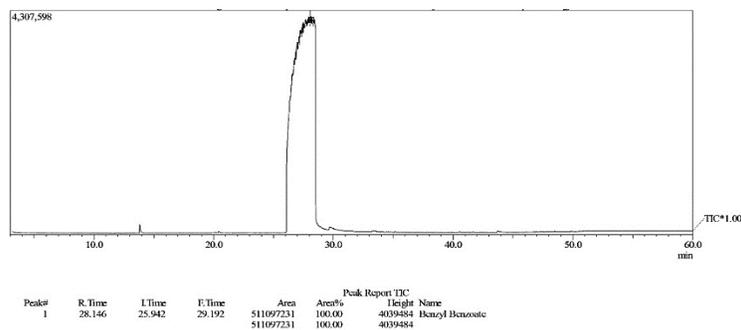


FIGURE 2. Chromatogram of benzyl benzoate

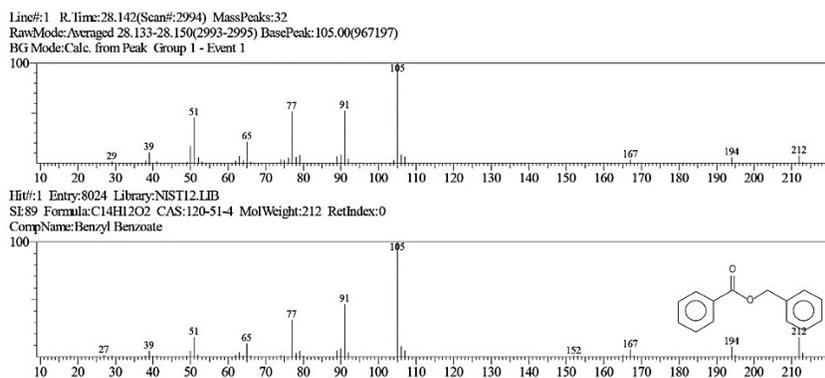


FIGURE 3. Mass spectrum of benzyl benzoate

The ^1H -NMR spectra (500 MHz, CDCl_3) of the isolated compound indicates eleven proton signals. They reveal an oxygenated methylene signal (δ 5.39, *s*), and ten signals of two aromatic rings. at δ 7.36 (1H, *t*, $J = 7.2$ Hz, H-3, H-5), δ 7.40 (2H, *t*, $J = 7.2$ Hz, H-3' and H-5'), δ 7.45 (1H, *t*, $J = 7.5$ Hz, H-4'), δ 7.47 (2H, *d*, $J = 7.2$ Hz, H-2', H-6'), δ 7.58 (1H, *t*, $J = 7.2$ Hz, H-4), and δ 8.10 (2H, *d*, $J = 7.5$ Hz, H-2, H-6) ppm. Three proton triplet and two proton doublet could be assigned to a phenyls group which indicates a substituent on an aromatic ring. The ^{13}C -NMR spectra confirms that there are 14 carbon signals, which indicated the presence of two sp^3 -carbon of oxygenated methylene ($\text{CH}_2\text{-O}$ -) at δ 66.70 (C-8) ppm, a sp^2 - carbon of carbonyl group at δ 166.43 (C-7) ppm, two substituted sp^2 -carbons at δ 130.15 (C-1) and 136.07 (C-1') ppm, and ten sp^2 -carbons bearing a hydrogen at δ 129.66 (C-2, C-6), δ 128.16 (C-3, C-5), δ 133.02 (C-4), δ 128.37 (C-2', C-6'), δ 128.40 (C-3', C-5') and δ 128.32 (C-4') ppm. The ^1H and ^{13}C NMR spectra data of benzyl benzoate was the newest data from previous data [16].

The carbon and proton correlation in one bond is determined by the 2D NMR spectra of HSQC (Heteronuclear Single Quantum Coherence), whereas the correlation of two or three bonds between carbon and proton is determined by the HMBC (Heteronuclear Multiple Bond Coherence) spectra. Table 1 showed the 1D (^1H and ^{13}C) and 2D (HSQC and HMBC) benzyl benzoate spectra data. Figure 4 describes the correlation proton and carbon (HMBC) of benzyl benzoate.

TABLE 1. The NMR spectra data of benzyl benzoate

C atom	HSQC		HMBC ($^1\text{H} \leftrightarrow ^{13}\text{C}$)
	δ_{C} ppm	δ_{H} (<i>mult</i> , <i>J</i> Hz) ppm	
1	130.15	-	-
2, 6	129.66	8,10 (2H, <i>d</i> , 7,3)	C-1, C-3, C-4, C-5, C-7
3, 5	128.16	7,36 (2H, <i>t</i> , 7,2)	C-2, C-4, C-7
4	133.02	7,58 (1H, <i>m</i> , 7,5)	C-2, C-3, C-5, C-6
7	166.43	-	-
8	66.70	5,38 (2H, <i>s</i>)	C-7, C-1', C-2'
1'	136.07	-	-
2', 6'	128.37	7,47 (2H, <i>d</i> , 7,2)	C-8, C-1', C-3', C-4'
3', 5'	128.40	7,40 (2H, <i>t</i> , 7,3)	C-1', C-2', C-4'
4'	128.32	7,45 (1H, <i>m</i> , 7,5)	C-2', C-3', C-5', C-6'

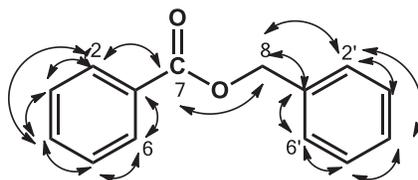


FIGURE 4. HMBC of benzyl benzoate

Toxicity Assays

Brine shrimp lethality test is a convenient method for monitoring bioactivities of various plant species. Although this method has not provided any adequate information about the mechanism of toxic action, it is a very useful method for the evaluation of the toxic potential of various plant extracts or isolated compounds [14].

Toxicity assay with the BSLT method was performed on acetone extract, n-hexane fraction and benzyl benzoate. The toxicity of samples that stated with LC_{50} values was compared with Meyer's or to Clarkson's toxicity index. According to Meyer's toxicity index, sample with $\text{LC}_{50} < 1000$ $\mu\text{g}/\text{mL}$ are considered as toxic, while extracts with $\text{LC}_{50} > 1000$ $\mu\text{g}/\text{mL}$ are considered as non-toxic [15]. Clarkson's toxicity criterion for the toxicity assessment of samples classifies them in the following order: samples with LC_{50} above 1000 $\mu\text{g}/\text{mL}$ are non-toxic, LC_{50} of 500 - 1000 $\mu\text{g}/\text{mL}$ is low toxic, samples with LC_{50} of 100 - 500 $\mu\text{g}/\text{mL}$ are medium toxic, and samples with LC_{50} of 0 - 100 $\mu\text{g}/\text{mL}$ are highly toxic [17].

TABLE 2. Toxicity of extract, fraction, and benzyl benzoate from *K. rotunda* L. rhizome

Sample	LC ₅₀ (µg/mL)
Acetone extract	35.86
n-Hexane fraction	49.80
Benzyl benzoate	173.49

Based on the Table 2 data, the LC₅₀ of benzyl benzoate most highest than acetone extract and *n*-hexane fraction this means the benzyl benzoate has lowest toxicity. The acetone extract and *n*-hexane fraction contain many compounds so that the synergistic effect of their compounds lead to higher activity of the pure compound. Synergistic effect is the interaction or combination of two or more compounds that produce the greater effects [18].

CONCLUSION

Toxicity assay of benzyl benzoate which isolated from the rhizome of *K. rotunda* L, using brine shrimp lethality test exhibited medium toxic with LC₅₀ value of 173.49 µg/mL. It was suggested that benzyl benzoate had a potential application to be developed as active cytotoxic and antitumor agents.

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Synthesis of salicylic acid modified magnetite nanoparticles and its application in wastewater treatment

Thutug Rahardiant Primadi, Fauziatul Fajaroh, Syaiful Bahri, Nazriati, Aman Santoso,

Endang Giyanti and Adhian Nugroho

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Modification of synthetic carpet using chitosan-titania nanocomposite for anti-bacterial and anti-odor purposes

Mohamad Iman Sulaeman, M. Ibadurrohman and Slamet

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Development of nanofluid biodegradable detergent from palm kernel oil and TiO₂

Reysa Anggraini Vestiana Putri, Muhammad Ibadurrohman and Slamet

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Preparation of activated carbon from *Calophyllum inophyllum* seed using different activating agents: Comparison study



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Nur Izzati Machrita, Kartika A. Madurani, Suprpto, M. Luki Kurniawan, Yulianto Adi Nugroho and Fredy Kurniawan

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Synthesis and characterization of tetrasulfapyridine-copper(II) sulfate trihydrate

Sentot Budi Rahardjo, Husna Syaima, Yuniar Dwi Andrieza, Witri Wahyu Lestari and Abu Masykur

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Preparation of starch-graft-acrylic acid/bentonite composite gel

Kaeksi Sekar Arum, Enggar Candra Prastiti, Prida Novarita Trisanti and Sumarno

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Identification of natural product compounds as NS5 RDRP inhibitor for dengue virus serotype 1-4 through in silico analysis

Hersal Hermana Putra, Mutiara Saragih, Yulianti and Usman Sumo Friend Tambunan

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Flexible molecular docking simulation of peptide compounds as inhibitor of Glul host protein for dengue fever therapy

Filia Stephanie, Ahmad Husein Alkaff and Usman Sumo Friend Tambunan

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The synthesis of surfactant by alcoholysis between glyceryl trilaurate and n-amyl alcohol

Daniel

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Study on the ion-exchange properties of the activated carbon black nanoparticles of ACBNPs20_17 code using sodium hydroxide solution

Pratama Jujur Wibawa, Muhammad Nur, Muhammad Asy'ari, Hadi Nur, Mohd. Arif Agam and Hashim Saim

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The effect of zeolite addition and freeze-drying method on alginat beads for controlled release fertilizer

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Chemical composition and antioxidant activities of citronella essential oil *Cymbopogon nardus* (L.) rendle fractions

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Initial study on the synthesis of 1-(4'-isopropilbenzil)-1,10-phenanthroline bromide from cuminyl alcohol, a potent antimalarial

Maulidan Firdaus, Soerya Dewi Marliyana and Muhammad Fajar Razak

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Freundlich adsorption isotherm in the perspective of chemical kinetics (II); rate law approach

Patiha, Maulidan Firdaus, Fitria Rahmawati, Sayekti Wahyuningsih and Triana Kusumaningsih

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Synthesis and spectra study of Cu (II), Fe (II), Zn (II)-5,15-diphenyl porphyrin

Atmanto Heru Wibowo, Metin Yuliati, Abu Masykur, Suyitno, Desi Suci Handayani, Dian Maruto Widjonarko, Maulidan Firdaus, Ari Yustisia and Takuji Ogawa

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Forward osmosis membrane to produce water energy drink from seawater

Saiful, Aida Afriyanti, Marlina, Muliadi Ramli and Abu Masykur

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Reusability study of fenton catalyst@bacterial celluloses for removal of methylene blue as synthetic dyes model

Husaini Ardy, Fakhri Arsyi Hawari, Ade Wahyu Y. P. Parmita, Untung Triadi, Azhar Isti Hanifah and Arie Wibowo

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Effect of phosphate ion on sorption of Nd(III) ion from aqueous solution using ion imprinted polymers

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Aging resistance and functional group analysis of natural rubber/oil palm empty fruit bunch charcoal composites

Hari Adi Prasetya, Popy Marlina and Rochmi Widjajanti

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Determination of the optimum composition to produce minimum particle size of β -carotene microencapsulated in acid hydrolyzed starch-chitosan/TPP (tripolyphosphate) matrices using Taguchi method

Agnes Dyah Novitasari Lestari, Mudasir, Dwi Siswanta and Ronny Martien

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The effect of coconut shell activated charcoal on vulcanization and morphology behaviour in natural rubber starch modified

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Synthesis and characterization of chitosan based super absorbent polymer modified with acrylic acid and acrylonitrile for Pb (II) metal ions removal from water

F. Widhi Mahatmanti, Harjono and Izzatun Niswah Assa'idah

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Hybrid PVA/alginate for extended delivery of antibiotic

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Synthesis of N'-(3-trimethoxysilylpropyl)diethylentriamine modified silica (SiO₂(RHA)-TMPDT) for adsorption of gold(III)



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Methyl red dye-sensitized zinc oxide as photocatalyst for phenol degradation under visible light

Wynona A. Nimpoeno, Hendrik O. Lintang and Leny Yuliati

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Crystalline carbon nitride for photocatalytic phenol degradation: Effect of precursor and salt melt amounts

Leny Yuliati, Mohd Hayrie Mohd Hatta, Siew Ling Lee and Hendrik O. Lintang

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Synthesis of CuO-TiO₂ nano-composite for *Escherichia coli* disinfection and toluene degradation



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Adsorption of Au(III) on diethylenetriamine-functionalized silica coated on iron sand magnetic material

Fahmiati, Alrum Armid, Suyanta and Nuryono

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Decolourization of methylene blue by NiO/ZSM-5 photocatalyst under UV-LED irradiation

Garcelina Rizky Anindika, Yuly Kusumawati, Didik Prasetyoko, Wahyu Bambang Widayatno and Abdul Hamid

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Isolation, characterization, and identification of endophytic



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leaves of *carica papaya* and its potential as an antioxidant

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Properties of starch biofoam reinforced with microcrystalline cellulose from banana stem fiber

Syahrul Fatrozi, Linda Purwanti, Sandra Kartika Sari, Muhammad Naufal Ariesta and Soerya Dewi Marliyana

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Analysis of chemical profile and antibacterial activity of secondary metabolites of endophytic fungi from *Annona squamosa* L. from Timor Island-Eastern Indonesia

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Gold (Au) selective adsorption using polyeugenol based ionic imprinted polymer with ethylene glycol dimethacrylate crosslink

M. Cholid Djunaidi, Nor Basid Adiwibawa Prasetya, Didik Setiyo Widodo, Retno Ariadi Lusiana and Pardoyo

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Synthesis of molecularly imprinted polymer urea based on polyeugenol with ethylene glycol dimethacrylate as crosslinking agent

M. Cholid Djunaidi, Arifatul Azizah and Gunawan

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The comparison of nitroxide radical derivative compound interaction with brookite and anatase surface: A guide to choose the best photoanode for DSSC application

Yuly Kusumawati, Leli D. Astuti, Eko Santoso and Syafsir Akhlus

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***In-vivo* acute toxicological studies of *Vasconcellea pubescens* A. DC. fruit extract against hepatic injury**

Heru Sasongko, Arifin Wicaksono and Sugiyarto

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Optimization of Suweg starch (*Amorphophallus paeoniifolius* (Dennst.) Nicolson) and lactose as *co-processed excipient* of Ibuprofen-PEG 6000 solid dispersion of effervescent tablet

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Developing formula of SNEDDS (*self nano emulsifying drug delivery system*) antihypertensive herbals “Hortus Medicus”

Dian Eka Ermawati, Roro Karina Pambudi, Vinda Aviwiandari, Yeni Farida, Sholichah Rohmani, Wisnu Kundarto and Estu Retnaningtyas Nugraheni

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Optimization of hydroxymethylcellulose and sodium CMC of transdermal patch of antihypertension “Hortus Medicus” and transport through membrane using franz diffusion cell method

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Liposomes from jack beans phospholipid extract for delivering vitamin C

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The effect of methyltriethoxysilane (MTES) concentration on hydrophobic properties of silica thin layer

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Synthesis zeolite y from kaolin: Activation of metakaolin with various concentration of sulfuric acid and its application for esterification

Leli Endah Safitri, Ulul Khairi Zuryati, Hannis Nur Rohma, Yatim Lailun Ni'mah and Didik Prasetyoko

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Synthesis of phenylcalix[4]resorcinarene sulfonate and its application as an antioxidant

Santi Nur Handayani, Heny Ekowati, Irmanto, Della Nadya Ayu Aprilia and Silva Utami

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The electronic properties study of betanin and their derivatives compound: An explanation to betanin limitation in DSSC application

Zulfa H. Damayanti, Garcelina R. Anindika, Eko Santoso, Syafsir Akhlus and Yuly Kusumawati

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Anthocyanin from butterfly pea flowers (*Clitoria ternatea*) by ultrasonic-assisted extraction

Achmad Qodim Syafa'atullah, Arie Amira, Sonya Hidayati and Mahfud Mahfud

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Synthesis and characterization of carbonaceous-based nanomaterials produced in chemical vapor deposition (CVD) using copper catalyst

Teguh Endah Saraswati, Ayu Dwi Priyanti, and Oktaviana Dewi Indah Prasiwi

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Preparation of NaFeO₂ from iron sand as a raw material for cathode of sodium-ion battery

Fitria Rahmawati, Arum A. Kusumaningtyas, Teguh E. Saraswati, Iwan Yahya and Younki Lee

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Chemical interaction analysis of L-Theanine compounds from *Camellia sinensis* L. with kainate glutamate receptors and their toxicity effect as anti autism candidates based on in silico

Mohamad Amin, Nanda Hilda Khikmawati, Suryadi, Ihya Fakhurizal Amin, Kodama Yayoi, Atmanto Heru Wibowo, Dina Maulina and Indriyani Rachman

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Synthesis, anticancer activity, and apoptosis mechanism of some chalcone derivatives

Hery Suwito, Helda Dwi Hardiyanti, Kautsar ul Haq, Alfinda Novi Kristanti, Umrotul Furghoniyyah, Aprillia Noni Rahmawati and Diwyareta Ristya Ayuningtyas

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Synthesis of 5-benzylidene-hydantoin and 5-benzylidene-creatinine derivatives under mixed catalyst systems of urea-*p*-toluenesulfonic acid (Urea-PTSA) and guanidine hydrochloride-triethylamine (GnHCl-TEA)

Kautsar Ul Haq, Septi Rosiana Dewi, Sherly Dwi Cicilianingrum, Amalia Muti Anggraini, Zella Dwipuspita Dahana, Indrianti Yunita Sari, Rina Dewi Renjanawati, Januardi Wardana, Fandi Gunawan, Nuzilatul Muschafi, Nisa'ur Rosyidah and Hery Suwito

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The compounds of styrene-butadiene rubber in the incorporation of palmitamide: Abrasion resistance, cure rate index and torque properties

Indra Surya and Edwin

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The compounds of montmorillonite-filled natural rubber: Cure rate index, swelling and hardness properties

I. Surya and H. Khosman

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Effect of low molecular weight organic acid (LMWOA) on the Zn²⁺ desorption from the soil of illegal land fill in Yogyakarta-Indonesia

Suherman, Ayu Maulidya Rachmanda, Roto and Kinichi Morita

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Microbial life on the surface of the soft coral for solve the self-healing concrete

Prima Endang Susilowati, Ahmad Zaeni, Sapril Kartini and I. Nyoman Sudiana

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Toxicity of benzyl benzoate from *Kaempferia rotunda* L. rhizome

Hartiwi Diastuti, Ari Asnani, Undri Rastuti and Mela Anggraeni

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Physico-chemical characteristics of gelatin as green template for nanomaterial production

Maria Ulfa and Windi Apriliani

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Intermolecular hydrogen bond interactions in *N*-carboxymethyl chitosan and *n*H₂O: DFT and NBO studies

Beti Safitri, Dwi Hudyanti, Marlyn Dian Laksitorini, Nurwarrohman Andre Sasongko and Parsaoran Siahaan

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Synthesis and anticancer study of complex nickel (II) 5,7-dibromoisatin-derived hydrazine carbothiamide

Fahimah Martak, Nofri Eka Safitri, Endah Mutiara Marhaeni Putri, Agung Bagus Pambudi and Arif Fadlan

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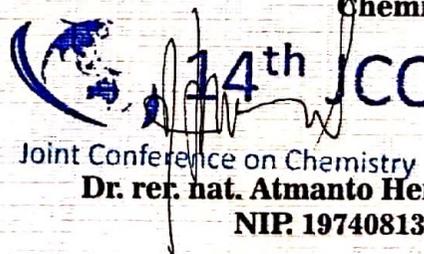
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